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## THE BUCCAL ABSORPTION OF ASCORBIC ACID USING CENTRIFUGED & UNCENTRIFUGED SALIVA

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### ABSTRACT

Vitamin C, also referred to as ascorbic acid (AA) or ascorbate, belongs to the water-soluble class of vitamins. Humans are one of the few species who lack the enzyme to convert glucose to vitamin C. Transport of vitamin C is a saturable and dose dependent process that occurs by active transport. At the intestine and cells AA is oxidized to Dihydro Ascorbic Acid (DHAA), which is more quickly transported across the cell membrane. Here we have reported the absorption of AA in the centrifuged and uncentrifuged saliva from the normal young adult man. The standard curve was prepared with different concentration ranging from 2-10 mcg/ml. The percentage of AA absorption was to be 93.85% in the uncentrifuged saliva which was significantly higher than the centrifuged saliva. AA absorption into the buccal mucosa was held in the mouth for periods of 1-9 minutes. For a constant mouth contact time, increase in pH of the loading solutions resulted in reduction of buccal absorption, and diminishing transfer into the buccal epithelium. The degree of intestinal absorption decreases as intake of AA increases.

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### Key Words

ascorbic acid Buccal absorption;  
Oral cavity; Buccal cavity;  
Perfusion cell; Absorption  
kinetics

## INTRODUCTION

Vitamin C, also known as L-ascorbic acid, is a water-soluble vitamin that is naturally present in some foods, added to others, and available as a dietary supplement. Humans, unlike most animals, are unable to synthesize vitamin C endogenously, so it is an essential dietary component [1].

Vitamin C is required for the biosynthesis of collagen, L-carnitine, and certain neurotransmitters; vitamin C is also involved in protein metabolism [1,2]. Collagen is an essential component of connective tissue, which plays a vital role in wound healing. Vitamin C is also an important physiological antioxidant [3] and has been shown to regenerate other antioxidants within the body, including alpha-tocopherol (vitamin E) [4]. Ongoing research is examining whether vitamin C, by limiting the damaging effects of free radicals through its antioxidant activity, might help prevent or delay the development of certain cancers, cardiovascular disease, and other diseases in which oxidative stress plays a causal role. In addition to its biosynthetic and antioxidant functions, vitamin C plays an important role in immune function [4] and improves the absorption of nonheme iron [5], the form of iron present in plant-based foods. Insufficient vitamin C intake causes scurvy, which is characterized by fatigue or lassitude, widespread connective tissue weakness, and capillary fragility [1,2,4,6-9].

The intestinal absorption of vitamin C is regulated by at least one specific dose-dependent, active transporter [4]. Cells accumulate vitamin C via a second specific transport protein. In vitro studies have found that oxidized vitamin C, or dehydroascorbic acid, enters cells via some facilitated glucose transporters and is then reduced internally to ascorbic acid. The physiologic importance of dehydroascorbic acid uptake and its contribution to overall vitamin C economy is unknown. Oral vitamin C produces tissue and plasma concentrations that the body tightly controls.

Approximately 70%–90% of vitamin C is absorbed at moderate intakes of 30–180 mg/day. However, at doses above 1 g/day, absorption falls to less than 50% and absorbed, unmetabolized ascorbic acid is excreted in the urine [4]. Results from pharmacokinetic studies indicate that oral doses of 1.25 g/day ascorbic acid produce mean peak plasma vitamin C concentrations of 135 micromol/L, which are about two times higher than those produced by consuming 200–300 mg/day ascorbic acid from vitamin C-rich foods [10]. Pharmacokinetic modeling predicts that even doses as high as 3 g ascorbic acid taken every 4 hours would produce peak plasma concentrations of only 220 micromol/L [10]. The total body content of vitamin C ranges from 300 mg (at near scurvy) to about 2 g [4]. High levels of vitamin C (millimolar concentrations) are maintained in cells and tissues, and are highest in leukocytes (white blood cells), eyes, adrenal glands, pituitary gland, and brain. Relatively low levels of vitamin C (micromolar concentrations) are found in extracellular fluids, such as plasma, red blood cells, and saliva [4].

### Recommended Intakes

- Recommended Dietary Allowance (RDA): average daily level of intake sufficient to meet the nutrient requirements of nearly all (97%–98%) healthy individuals.
- Adequate Intake (AI): established when evidence is insufficient to develop an RDA and is set at a level assumed to ensure nutritional adequacy.
- Tolerable Upper Intake Level (UL): maximum daily intake unlikely to cause adverse health effects [8].

Table 1 lists the current RDAs for vitamin C [8]. The RDAs for vitamin C are based on its known physiological and antioxidant functions in white blood cells and are much higher than the amount required for protection from deficiency [4,8,11]. For infants from birth to 12 months, the FNB established an AI for vitamin C that is equivalent to the mean intake of vitamin C in healthy, breastfed infants.

**Table 1:** Recommended Dietary Allowances (RDAs) for Vitamin C [8]

Age	Male	Female	Pregnancy	Lactation
0–6 months	40 mg*	40 mg*		
7–12 months	50 mg*	50 mg*		
1–3 years	15 mg	15 mg		
4–8 years	25 mg	25 mg		
9–13 years	45 mg	45 mg		
14–18 years	75 mg	65 mg	80 mg	115 mg
19+ years	90 mg	75 mg	85 mg	120 mg
Smokers	Individuals who smoke require 35 mg/day more vitamin C than nonsmokers.			

**MATERIAL AND METHOD****MATERIAL;**

Ascorbic acid, H<sub>2</sub>SO<sub>4</sub>, Ammonium Molybdate and Distilled Water was obtained from well reputed research laboratory and U.V.Spectrophotometer.

**METHOD-**

100mg of Ascorbic acid was weighed and taken into 100ml of volumetric flask and diluted with distilled water till the mark.10 ml solution was taken from above solution and diluted to 100ml with distilled water(100mg/ml).Aliquot were prepared by using 0.5ml,1.0ml,2.0ml,3.0ml,4.0ml and 5.0ml of solution in 25.0ml volumetric flask.

2.0ml of sample+2.0ml H<sub>2</sub>SO<sub>4</sub>(10%w/v)+4ml of Ammonium Molybdate(10%w/v) solution kept at room temperature for 60 minutes. Absorbance was measured at 650nm.

**SAMPLE SOLUTION-**

Ascorbic acid 25ml solution,concentration 100mcg/ml was taken in a beaker and kept it in mouth and collected 35 times for 5min.Expelled the solution and collected in beaker,again mouth was rinsed with 10ml distilled water and expelled in beaker(volume become 35ml).2.0ml of above solution was taken and added 0.8ml H<sub>2</sub>SO<sub>4</sub>(5%w/v) and 1.0ml Ammonium Molybdate in 10.0ml of volumetric flask and volume was made up 10.0ml with distilled water.solution was kept in dark room.Absorbance was was measured at 650nm.

**EVALUATION****Absorption and body stores**

Intestinal absorption of vitamin C depends on the amount of dietary intake, decreasing with increasing intake levels. At an intake of 30 to 180 milligrams, about

70% to 90% is absorbed; about 50% of a single dose of 1 to 1.5 grams is absorbed; and only 16% of a single dose of 12 grams is absorbed. Up to about 500 milligrams are absorbed via a sodium-dependent active transport process, while at higher doses simple diffusion occurs.

The storage capacity of water-soluble vitamins is generally low compared to that of fat-soluble ones. Humans have an average tissue store of vitamin C of 20 mg/kg body weight. The highest concentration is found in the pituitary gland (400 mg/kg); other tissues of high concentration are the adrenal glands, liver, brain and white blood cells (leukocytes).

**Measurement**

Vitamin C can be measured in the blood plasma and other body tissues by various techniques. Also dipstick tests for estimation of vitamin C levels in the urine are available. Less satisfying, however, is the evaluation of the analytical data concerning the true reflection of the body status. Threshold values are difficult to define and the subject of controversial discussion. Typical blood plasma levels are in the range of 20 to 100 µmol/L

**Stability**

Vitamin C is sensitive to heat, light and oxygen. In food it can be partly or completely destroyed by long storage or overcooking. Refrigeration can substantially diminish vitamin C loss in food.

**Positive interactions**

The presence of other antioxidants, such as vitamin E and beta-carotene, supports the protective antioxidant action of vitamin C. Other vitamins, such as those of the B-complex (particularly B<sub>6</sub>, B<sub>12</sub>, folic acid and pantothenic acid) and some pharmacologically active substances, as well as the naturally occurring

compounds known as bioflavonoids, may have a sparing effect on vitamin C.

#### **Negative interactions**

Due to toxic compounds in smoke, the vitamin C requirement for smokers is about 35 mg/day higher than for non-smokers. Also several pharmacologically active compounds, among them some anti-depressants, diuretics, birth control pills and aspirin, deplete the tissues of vitamin C. This is also true of certain habits, for example alcohol consumption.

#### **Disease prevention and therapeutic use**

Dozens of prospective studies suggest that vitamin C plays a role in preventing a variety of diseases. It is also used to treat certain diseases in orthomolecular medicine. As this nutrient is important for a variety of diseases, only a selection of them are presented here in detail.

#### **Cardiovascular diseases (CVD) (heart disease and stroke)**

The data for the CVD protective benefits of vitamin C are inconsistent. While some studies have failed to find significant reductions in the risk of coronary heart disease (CHD), numerous prospective cohort studies have found inverse associations between dietary vitamin C intake or vitamin C plasma levels and CVD risk. Vitamin C may protect coronary arteries by reducing the build-up of plaque, as this helps to prevent the oxidation of LDL cholesterol (the “bad” cholesterol), especially in combination with vitamin E. Some data has shown that vitamin C may also boost blood levels of HDL cholesterol (the “good” cholesterol), which is also considered positive for the prevention of heart diseases. The risk of stroke may be reduced by an adequate intake of vitamin C through fruits, vegetables and supplements. However, due to the inconsistency of the data and its lack of specificity to vitamin C, the interpretation of these results is difficult.

#### **Cancer**

The role of vitamin C in cancer prevention has been studied extensively, and until now no beneficial effect has been shown for breast, prostate, or lung cancer. However, a number of studies have associated higher intakes of vitamin C with decreased incidence of cancers of the upper digestive tract, cervix, ovary, bladder, and

colon. Studies finding significant cancer risk reduction by dietary intake recommended at least 5 servings of fruits and vegetables per day. Five servings of most fruits and vegetables provide more than 200 mg vitamin C per day. Just significant cancer risk reductions were found in people consuming at least 80 to 110 mg of vitamin C daily.

#### **Common cold**

Numerous studies have shown a general lack of effect of prophylactic vitamin C supplementation on the incidence of common cold, but they do show a moderate benefit in terms of the duration and severity of episodes in some groups, especially those who are exposed to substantial physical and/or cold stress. The improvement in severity of colds after vitamin C supplementation may be due to the antihistaminic action of mega doses of vitamin C.

#### **Wound healing**

During a postoperative period, or during healing of superficial wounds, supplemental vitamin C contributes to the prevention of infections and promotes skin repair.

#### **Blood pressure**

Several studies have shown a blood pressure lowering effect of vitamin C supplementation at about 500 mg per day due to improved dilation of blood vessels.

#### **Recommended Dietary Allowance (RDA)**

The recommended daily intake of vitamin C varies according to age, sex, risk group and criteria applied in individual countries. The RDAs in the USA for vitamin C were recently revised upwards to 90 mg/day for men and 75 mg/day for women, based on pharmacokinetic data. For smokers, these RDAs are increased by an additional 35 mg/day. Higher amounts of vitamin C are also recommended for pregnant (85 mg/day) and lactating women (120 mg/day). The RDAs are in a similar range in other countries. Recent evidence sets the estimate for the maintenance of optimal health in the region of 100 mg daily.

#### **Safety**

As much as 6-10 g vitamin C per day (more than 100 times the RDA) has been ingested regularly by many people with no evidence of side effects. Although a number of possible problems with very large doses of

vitamin C have been suggested, none of these adverse health effects have been confirmed, and there is no reliable scientific evidence that large amounts of vitamin C (up to 10 g/day in adults) are toxic. In the year 2000 the US Food and Nutrition Board recommended a tolerable upper intake level (UL) for vitamin C of 2 g (2,000 mg) daily in order to prevent most adults from experiencing osmotic diarrhoea and gastrointestinal disturbances.

### Uses in food technology

The food industry uses ascorbic acid as a natural antioxidant. This means that ascorbic acid, added to foodstuffs during processing or prior to packing, preserves colour, aroma and nutrient content. This use of ascorbic acid has nothing to do with its vitamin action. In meat processing, ascorbic acid makes it possible to reduce both the amount of added nitrite and the residual nitrite content in the product. The addition of ascorbic acid to fresh flour improves its baking qualities, thus saving the 4-8 weeks of maturation flour would normally have to undergo after milling.

### RESULT AND DISCUSSION

Vitamin c is an essential vitamin for human life and it is most beneficial for human health in the formation of immunity. It work against the various micro-organisms and treatment of disease like cancer, blood pressure, wound healing, cardiovascular disease, common cold and prevalent various disease. It is also used as a nutrients. It is also act as a natural antioxidant. Absorption of ascorbic acid was found to be 93.85% in the uncentrifuged saliva which was significantly higher than the centrifuged saliva. Vitamin C absorption into the buccal mucosa was held in the mouth for periods of 1-9 minutes. For a constant mouth contact time, increase in pH of loading solutions resulted in reduction of buccal absorption, and diminishing transfer into the buccal epithelium. The degree of intestinal absorption decreases as intake of vitamin C increases.

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